

Rhenium ${}_{75}\text{Re}^{186.2}$

Discovered in 1925 by W. Noddack, Ida Tacke and O. Berg at Berlin, Germany.

[Greek, Rhenus = river Rhine]

French: Rhénium

German: Rhenium

Italian: Renio

Spanish: Renio



Atomic number	75
Density in g/cm ³	21.04
Atomic radius in pm	188
Atomic weight	186.21
Melting point in °C	3180
Boiling point in °C	5630

Description: Rhenium is a silvery metal, but is usually obtained as a grey powder. It resists corrosion but slowly tarnishes in moist air. Rhenium dissolves in HNO₃ and H₂SO₄. It is used in filaments, thermistors and catalysts.

RHENIUM SINGLE CRYSTAL PROPERTIES

State:	Single crystal
Crystal structure:	hexagonal
Production method:	Floating zone
Standard size:	diameter 6mm thickness 1-2mm
Orientation:	(0001), (1-100) and (11-20)
Orientation accuracy:	<2°, <1°, <0.4° or <0.1°
Polishing:	as cut, one or two sides polished
Roughness of surface:	<0.03µm
Purity:	99.99%
	C 3
	H < 1
	O 9
	N < 5
	Cu 1.60
	Fe 1.80
	Ni < 1
	Pb 0.30
	Si 0.30
	Ga, Hf and Ta are below the detection limit
Density:	12.4 g/cm ³
Melting point:	3179.85 °C / 3453 °K
Boiling point:	5626.85 °C / 5900 °K
Molar volume:	8.86 cm ³
Thermal conductivity:	47.9 [300 K] Wm ⁻¹ K ⁻¹
Coefficient of linear thermal expansion:	6.63 x 10 ⁻⁶ K ⁻¹
Typical analysis (ppm):	

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Electrical resistivity:	19.3x 10 ⁻⁸ [293 K] Wm
Mass magnetic susceptibility:	+4.56 x 10 ⁻⁹ (s) kg-lm ³
Young's modulus:	466 GPa
Rigidity modulus:	181 GPa
Bulk modulus:	334 GPa
Poisson's ratio:	0.26
Radii:	Re7+ 60; Re6+ 61; Re4+ 72; atomic 137; covalent 12
Electronegativity:	1.9 (Pauling); 1.46 (Allred); 4.02 eV (absolute)
Effective nuclear charge:	3.60 (Slater); 10.12 (Clementi); 14.62 (Froese-Fischer)
Number of Isotopes (incl. nuclear isomers):	34
Isotope mass range:	162 -> 192
Crystal structure, (cell dimensions / pm), space group	hexagonal
X-ray diffraction: mass absorption coefficients:	CuK α 179 (μ/r) / cm ² g ⁻¹ MoK α 103 (μ/r) / cm ² g ⁻¹
Neutron scattering length:	0.92 b/10 ⁻¹² cm
Thermal neutron capture cross-section:	89.7 sa / barns

